# nature communications

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# Imaging 3D chemistry at 1 nm resolution with fused multi-modal electron tomography

Received: 14 December 2023

Accepted: 3 April 2024

Published online: 26 April 2024

Check for updates

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Measuring the three-dimensional (3D) distribution of chemistry in nanoscale matter is a longstanding challenge for metrological science. The inelastic scattering events required for 3D chemical imaging are too rare, requiring high beam exposure that destroys the specimen before an experiment is completed. Even larger doses are required to achieve high resolution. Thus, chemical mapping in 3D has been unachievable except at lower resolution with the most radiation-hard materials. Here, high-resolution 3D chemical imaging is achieved near or below one-nanometer resolution in an Au-Fe<sub>3</sub>O<sub>4</sub> metamaterial within an organic ligand matrix, Co<sub>3</sub>O<sub>4</sub>-Mn<sub>3</sub>O<sub>4</sub> core-shell nanocrystals, and ZnS-Cu<sub>0.64</sub>S<sub>0.36</sub> nanomaterial using fused multi-modal electron tomography. Multi-modal data fusion enables high-resolution chemical tomography often with 99% less dose by linking information encoded within both elastic (HAADF) and inelastic (EDX/EELS) signals. We thus demonstrate that sub-nanometer 3D resolution of chemistry is measurable for a broad class of geometrically and compositionally complex materials.

Knowing the complete chemical arrangement of matter in all dimensions is fundamental to engineering novel nanomaterials<sup>1</sup>. Although electron tomography provides comprehensive 3D structure at resolutions below 1 nm using elastic scattering signals<sup>2–4</sup>, chemical tomography obtained from inelastic scattering remains largely out of reach. Several demonstrations of chemical tomography using electron energy loss or x-ray energy spectroscopy (EELS/EDX) accompanied the introduction of scanning transmission electron microscope (STEM) tomography and provide a milestone for 3D imaging<sup>5-8</sup>. However, chemical tomography from core-excitation spectroscopy demands high electron doses that almost always exceed the specimen limits (e.g., >10<sup>7</sup> e/Å<sup>2</sup>)<sup>9-11</sup>. If attempting chemical tomography, researchers

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**Fig. 1** | **Nanoscale recovery of Au-Fe<sub>3</sub>O<sub>4</sub> nanoparticle superlattice.** a Schematic highlighting the linked HAADF and EELS modalities for chemical tomography<sup>84</sup>. HAADF projection images are collected at every tilt increment while core-loss EELS spectra are sparsely acquired every few tilts. **b** The fused multi-modal

reconstruction for the specimen's Fe  $L_{2,3}$  (turquoise), O-K (turquoise), and gold  $M_{4,5}$  edge (yellow).  $\bm{c}$  Chemical overlay of the superlattice nanoparticles over the entire 115 nm field of view. Scale cubes, 5 nm<sup>3</sup>.

must sacrifice resolution by collecting few specimen projections (e.g., 5-10) and constrain the total dose (e.g.,  $<10^{6} \text{e/Å}^{2}$ ). Consequently, 3D resolution is penalized from undersampling and noisy chemical maps<sup>12</sup>. Therefore, a paradigm shift is necessary for high-resolution chemical tomography.

Achieving high-resolution 3D chemistry at lower dose requires fusing both elastic and inelastic scattering signals. Typically these detector signals are analyzed separately and correlated<sup>13-18</sup>. However, correlative imaging disregards shared but also complementary information between structure and chemistry and misses opportunities to recover useful information<sup>19</sup>. Data fusion, popularized in satellite imaging, goes further than correlation by linking separate signal modalities to reconstruct new information and improve measurement accuracy<sup>20-22</sup>. Early approaches for tomographic data fusion showed noise reduction through joint regularization across modalities<sup>23</sup> and linear models between EDX and elastic signals<sup>24-26</sup>. Recent developments in 2D multi-modal data fusion integrate the non-linear physics of Rutherford scattering to substantially reduce the dose requirements to acquire an atomic-resolution map<sup>27</sup>. In alignment with this principle of fused multi-modal electron microscopy, we extend its algorithmic framework into the third dimension.

Here we present fused multi-modal electron tomography that offers high-resolution recovery of nanomaterial chemistry in 3D with high signal-to-noise (SNR) by fusing signals from both elastic high-angle annular dark field (HAADF) and inelastic (EDX/EELS) scattering. Multi-modal electron tomography reconstructs the volumetric chemical structure of specimens by solving a threeterm inverse problem that fuses information from multiple detectors. This approach extends beyond 2D data fusion<sup>27</sup> to offer a 3D framework with distinct sampling strategies that minimize dose and maximize resolution. When many HAADF projections are measured alongside far fewer chemical projections 100-fold dose reductions are achievable. Although the 3D chemical structure is severely underdetermined, fusing both modalities allows missing chemical information to become identifiable. This approach demonstrates that researchers can measure 3D chemistry at 1 nm resolution using electron doses as low as  $10^4 \text{ e/}\text{\AA}^2$  and as few as nine spectroscopic maps while remaining consistent with original measurements. Multi-modal tomography is validated across multiple material systems, including Au-Fe<sub>3</sub>O<sub>4</sub> superlattice clusters, core-shell Co<sub>3</sub>O<sub>4</sub>-Mn<sub>3</sub>O<sub>4</sub><sup>28</sup>, ZnS-Cu<sub>0.64</sub>S<sub>0.36</sub> heterostructures<sup>29</sup>, Cu-SiC nanoparticles and several simulated specimens. By fusing modalities, chemical tomography is possible at sub-nanometer resolution along all three dimensions is achievable for a wider class of material systems.

#### Results

#### Principles of fused multi-modal electron tomography

High-resolution 3D chemical imaging is achieved using the multimodal electron tomography framework illustrated in Fig. 1a for a binary Au-Fe<sub>3</sub>O<sub>4</sub> nanoparticle superlattice grafted with thiol endfunctionalized polystyrene ligands. In multi-modal electron tomography, projections of the specimen structure are measured from a HAADF detector and the specimen chemistry is extracted from spectroscopy (EELS or EDX). These two detector modalities are fused during the reconstruction process to provide the complete 3D chemical distribution of a specimen at high resolution and SNR. Figure 1b shows the 3D reconstruction of each individual chemistry: larger  $10.2 \pm 1.1$  nm Fe nanoparticles (blue) and smaller Au  $3.9 \pm 0.4$  nm nanoparticles (orange). Both chemistries are visualized simultaneously in Fig 1c to show the self-organization of the chemical superlattice. The light-element, carbon matrix is shown in Supplementary Fig. 1.

In multi-modal tomography, the number of structural HAADF projections usually exceeds the chemical projections. In this first demonstration, only nine chemical maps ( $\Delta\theta$  = 15°) are measured from the Fe-L<sub>2,3</sub> and Au-M<sub>4,5</sub> core-excitation edges in an EELS spectrum whereas 47 HAADF images ( $\Delta\theta$  = 3°) are collected over a ± 70° specimen tilt range. Linking both modalities into the reconstruction enables a clear distinction between Fe<sub>3</sub>O<sub>4</sub> and Au nanoparticles at high resolution from just a few EELS maps and a total electron dose of 5 × 10<sup>5</sup> e/Å<sup>2</sup>-roughly two orders of magnitude lower total electron dose than an equivalent conventional approach.

Fused multi-modal electron tomography reconstructs threedimensional chemical models by solving an optimization problem seeking a solution that strongly agrees with (1) the HAADF modality containing high SNR, (2) the chemically sensitive spectroscopic modality (EELS and/or EDX), and (3) encourages sparsity in the gradient domain producing solutions with reduced spatial variation. The overall



Fig. 2 | Nanoscale recovery of  $Co_3O_4$ -Mn<sub>3</sub>O<sub>4</sub> core-shell nanoparticles. a–c Raw EELS reconstruction for the Co (blue-green) and Mn (orange) L<sub>2,3</sub> core-loss edges. d–f The HAADF tomogram of Co<sub>3</sub>O<sub>4</sub>-Mn<sub>3</sub>O<sub>4</sub> nanoparticle tracks the structure of the specimen but fails to describe materials chemistry in 3D. g–i The fused multi-modal

reconstruction. Scale cubes, 25 nm<sup>3</sup>. **a**, **d**, **g** Representation in Fourier space of the projections used to reconstruct the tomograms. **j** Fused multi-modal tomogram of a single  $Co_3O_4$ -Mn<sub>3</sub> $O_4$  nanoparticle. Scale cube, 10 nm<sup>3</sup>. **k** A line profile showing the average intensity across the diameter of the particle.

optimization function is as follows:

$$\arg\min_{\boldsymbol{x}_{i} \geq 0} \quad \frac{\lambda_{1}}{2} \left\| \boldsymbol{A}_{h} \sum_{i} (Z_{i} \boldsymbol{x}_{i})^{\gamma} - \boldsymbol{b}_{H} \right\|_{2}^{2} + \\ \lambda_{2} \sum_{i} \left( \boldsymbol{I}^{T} \boldsymbol{A}_{c} \boldsymbol{x}_{i} - \boldsymbol{b}_{i}^{T} \log(\boldsymbol{A}_{c} \boldsymbol{x}_{i} + \varepsilon) \right) + \lambda_{3} \sum_{i} \|\boldsymbol{x}_{i}\|_{\mathrm{TV}},$$
(1)

 $\mathbf{x}_i$  is the reconstructed 3D chemical distributions for element *i*,  $\mathbf{b}_i$  is the measured 2D chemical maps for element *i*,  $\mathbf{b}_H$  is the measured HAADF micrographs,  $\mathbf{A}_h$  and  $\mathbf{A}_c$  are forward projection operators for HAADF and chemical modalities,  $\lambda$  are regularization parameters,  $\varepsilon$  herein prevents log(0) issues but can also account for background, the log is applied element-wise to its arguments, superscript *T* denotes vector transpose, and *I* denotes the vector of  $N_{\text{chem}}^{\text{proj}} n_y n_i$  ones, where  $n_y$  is the number of pixels,  $n_i$  is the number of elements present, and  $N_{\text{chem}}^{\text{proj}}$  is the number of projections for the chemical modality. Pseudo-code for numerical implementation is provided in the Supplementary Materials.

The three terms in Eq. (1) define our fused multi-modal framework designed to surpass traditional limits for chemical tomography. First, we assume a forward model where the simultaneous HAADF is a linear combination of the reconstructed 3D elemental distributions ( $\mathbf{x}_i^{\gamma}$  where  $\gamma \in [1.4, 2]$ ). The incoherent linear imaging approximation for elastic scattering scales with atomic number as  $Z_i^{\gamma}$ , where experimentally  $\gamma$  is typically around  $1.7^{30-32}$ . This  $\gamma$  is bounded between 4 and 3 as described by Lenz–Wentzel expressions for electrons passing through a screened coulombic potential and 2 for Rutherford scattering from

bare nuclear potentials<sup>33,34</sup>. Second, we ensure the recovered 3D distributions maintain a high degree of data fidelity with the initial measurements by using the log-likelihood for spectroscopic measurements dominated by low-count Poisson statistics<sup>22,35</sup>. In a higher count regime, this term can be substituted with a least-squares discrepancy ( $|| Ax - b||_2^{2}$ )<sup>36</sup>. Lastly, we include channel-wise isotropic total variation (TV) regularization to enforce a sparse gradient magnitude, which reduces noise by promoting image smoothness while preserving sharp features<sup>37</sup>. This sparsity constraint, popularized by the field of compressed sensing (CS), is a powerful yet modest prior for recovering structured data<sup>38,39</sup>. When solving Eq. (1), each of these three terms should be weighted appropriately by determining coefficients ( $\lambda$ ) that balance their contributions. Ultimately, optimization of all three terms is necessary for accurate recovery (Supplementary Figs. 2, 3).

The improvement in reconstruction quality with fused multimodal chemical tomography (Fig. 2i) is dramatic when compared to traditional chemical tomography (Fig. 2c).

#### 3D chemistry at high-resolution, low-dose

In tomography, 3D resolution is described by the Crowther criterion, which states resolution is limited by the object size and the number of specimen projections measured<sup>40</sup>-higher resolution requires more projections<sup>41</sup>. For traditional chemical tomography, few chemical projections are collected and the Crowther relation devastates resolution in 3D. This limitation occurs from the highdose requirements of chemical mapping (i.e., EDX, EELS) where only a few projections can be collected before radiation damage alters the specimen structure.

Figure 2 shows how specimen projections from each modality are superimposed as planes of information in Fourier space. Chemical tomography is sparsely sampled in Fourier space (Fig. 2a), which results in a tomographic reconstruction containing artifacts and low SNR (Fig. 2b, c). Despite the poor quality, traditional chemical tomography tracks the chemical distribution, and the Mn shell (orange) can be seen surrounding the Co core (blue-green). In contrast, elastically scattered electrons collected by the HAADF detector provide high signals at lower doses and allow many projections to be collected-in practice, HAADF sampling is five to ten times more finely spaced than chemical (Fig. 2d)<sup>32</sup>. The dose required for a single HAADF projection is 10<sup>2</sup>–10<sup>3</sup> times lower than a chemical projection acquired using coreenergy loss spectroscopy. Thus, it is favorable to acquire more HAADF images and achieve higher resolution. Although HAADF tomography permits high-resolution and high-SNR reconstructions of structure, it lacks chemical specificity. This is seen in Fig. 2e, f where the structure is well defined with low noise but the Co and Mn regions are not identifiable.

Exploiting shared information in both modalities, multimodal tomography achieves a chemical resolution in 3D comparable to high-resolution HAADF reconstructions. Although few chemical measurements pose a severely underdetermined problem, fusing with the HAADF modality fills in missing chemical information. This is reflected in Fig. 2g where many HAADF projections (e.g., 50–180) are measured while far fewer chemical projections (e.g., 5–15) are intermittently measured. In this reconstruction, 9 EELS maps and 45 HAADF projections (50–200 mrad detector inner and outer semi-angles) were collected over a  $\pm 60$ ° tilt range using a 2.4 Å probe with a 24.3 nm depth of focus (300 keV acceleration voltage, 10 mrad convergence angle). High-resolution 3D chemistry is visible in the core-shell Co<sub>3</sub>O<sub>4</sub>-Mn<sub>3</sub>O<sub>4</sub> using multi-modal tomography in Fig. 2h, i.

Fused multi-modal electron tomography provides unique insight for studying heterostructured nanocrystals with unprecedented geometries. In the case of Co<sub>3</sub>O<sub>4</sub>-Mn<sub>3</sub>O<sub>4</sub> nanocrystals, the manganese oxide shell is divided into several ordered grains that grow on each surface plane for the cobalt oxide nanocube core<sup>28</sup>. However, the core and shell interface can vary per plane driven by the growth interplay between strain and surface energy, resulting in the formation of grain boundaries<sup>42</sup>. The complete 3D distribution of Co and Mn at the surface and interface is difficult to discern with 2D projected EELS maps or HAADF reconstructions. Fortunately, the fused chemical distributions reveal surface coverage of the shell grains and cross-sections quantify the shell thickness and interface chemistry (Fig. 2k). To further demonstrate, fused multi-modal EELS tomography was used to discern between ZnS and Cu<sub>0.64</sub>S<sub>0.36</sub> phases (Supplementary Fig. 4) in a heterostructured nanocrystal<sup>29</sup> and EDX tomography to identify Cu nanoparticles embedded in SiC catalysts (Supplementary Fig. 5).

Data fusion eliminates noticeable noise in the final 3D chemical reconstruction without a loss of resolution. This noise reduction accompanies a dose reduction of roughly 100 fold. Linking the chemical projections to the high SNR HAADF signals dose-efficiently boosts the chemical specificity. Even at modest HAADF signals (e.g., SNR  $\simeq$  10), multi-modal tomography notably outperforms traditional chemical tomography (Supplementary Fig. 21). To illustrate, in Fig. 2, matching the resolution of fused multi-modal chemical tomography using traditional methods would require 45 EELS maps—a fivefold dose increase. However, the SNR of each chemical projection would still fall short (Supplementary Fig. 19) and require roughly 20-times additional dose. In total, multi-modal chemical tomography performs well at one-hundredth the dose requirement of traditional methods.

Reduction of electron beam dose produces irreplaceable advantages for electron tomography—both in terms of accessible resolution and the range of materials classes that can be imaged in 3D. Dose

#### Sub-nanometer chemical resolution in 3D

3D resolution of the chemical distribution in Au-Fe<sub>3</sub>O<sub>4</sub> nanoparticle superlattice (Fig. 3a) is demonstrated at or below 1 nm using multimodal tomography. The achieved resolution is quantified in real and reciprocal space. In real space, the resolution limit is verified by visually inspecting a single 3 nm Au nanoparticle (Fig. 3d). The edge sharpness between the reconstructed nanoparticle and vacuum is visibly less than 1 nm. From line profiles, the half-pitch resolution is 0.8 nm × 0.8 nm × 1.1 nm along the *x*, *y*, and *z* directions, respectively. Along optimal directions (*x*, *y*) the resolution is comparable to the Nyquist frequency (8.05 Å). The real-space resolution is consistent



**Fig. 3** | **Resolution analysis of Au-Fe<sub>3</sub>O<sub>4</sub> superlattice nanoparticles.** a Fused EELS tomograms of Au-Fe<sub>3</sub>O<sub>4</sub> nanoparticles. Scale cube, 2 nm<sup>3</sup>. **b** Power spectral density of the Fe reconstruction along the principal axial directions shown on the right. Scale bar,  $0.5 \text{ nm}^{-1}$ . **c** Power spectral density profiles for  $k_x k_y$  and  $k_x k_z$  directions. **d** A 2.5 Au nanoparticle is shown with, **e**, line profiles showing a resolution of 0.8 nm, 0.8 nm, and 1.1 nm along the x, y, and z directions. **f** Planar cross sections of the 2.5 nm. Au nanoparticle.



**Fig. 4** | **Estimating sampling requirements for accurate recovery with synthetic CoO/CuO nanocubes. a** An normalized root mean square error (NRMSE) map representing the reconstruction error as a function of the number of HAADF and chemical tilts. Brighter pixels denote results containing incorrect reconstructions from the ground truth. **b** Visualization of three points corresponding to conventional chemical tomography (reconstruction without the HAADF modality), and low or high-dose fused multi-modal electron tomography. **c** The 3D models used for generating synthetic chemical and ADF projections<sup>85</sup>. Scale bar, 75 nm.

with reciprocal space estimates of the cutoff frequency at which the signal drops to the noise floor<sup>1</sup>. Figure 3b highlights power spectral density variations projected on three orthogonal planes. Measured power spectral density along the  $k_x k_y$  and  $k_x k_z$  directions show information transfer roughly occurring at 0.99 nm and 1.02 nm respectively (Fig. 3c). These directions conservatively represent the 3D resolution from an average of the high-resolution and low-resolution (z-axis) directions. This 3D chemical resolution nearly matches the 3D HAADF resolution 1.00 nm, 1.01 nm in Fig. 3 (Supplementary Fig. 8).

For fused multi-modal chemical tomography, the HAADF 3D resolution provides an upper bound to the highest obtainable 3D chemical resolution. A reduction of resolution along the *z*-axis is expected from the incomplete tilt range that creates a missing wedge of information in Fourier space<sup>45</sup>. Avoiding this anisotropic resolution loss has been demonstrated by acquiring a full tilt range (±90°) through the preparation of needle wire samples or preparing nanoparticles on carbon nanofibers<sup>46,47</sup>. Here, we observe approximately a 25% reduction in resolution along the missing wedge direction of the multi-modal chemical reconstruction (Fig. 3e, f).

#### Influence of sampling

Electron tomography simulations show a 3-5 fold improvement in the normalized root mean square error ((NRMSE)) averaged across all elements when multi-modal tomography is used over conventional chemical tomography. In Fig. 4 synthetic gold decorated CoO/CuO nanocubes inspired by real experimental data<sup>47</sup> provide a ground truth comparison to assess the accuracy of fused multi-modal tomography. Simulated projection images are generated from a simple linear incoherent imaging model of the 3D chemical composition with added Poisson noise (See Methods). The specimen tilt range is limited to  $\pm 70^{\circ}$ to better match typical experimental conditions. The advantages of multi-modal tomography are clearly visible in the 2D slices (Fig. 4b) taken from 3D reconstructions obtained by conventional chemical tomography ( $\langle NRMSE \rangle = 1.301$ ) and fused multi-modal tomography  $(\langle NRMSE \rangle = 0.33)$ . For all chemistries (Au, O, Cu, Co) fused multimodal tomography is more consistent with the ground truth with higher resolution and reduced noise.

For any number of chemical projections acquired, we see a notable reduction in NRMSE when HAADF projections are integrated into the chemical reconstruction. Figure 4 shows the improved fused multi-modal reconstruction accuracy across a wide range of HAADF and chemical projections for the gold-decorated CoO/CuO nanocubes. The reconstruction error (average NRMSE) across most of the multi-modal parameter space is less than 0.40 compared to values around 1.2 for conventional tomography. Pixel values on the diagram (Fig. 4a) represent the average NRMSE across all elements. This NRMSE map shows data fusion strongly benefits by increasing the HAADF information available. It requires substantially less dose to increase the HAADF projections (i.e., moving vertically on the map) compared to increasing the chemical projections (i.e., moving horizontally on the map).

Conventional chemical tomography does not use HAADF projections (bottom row, Fig. 4a) resulting in an average reconstruction error larger than the entire multi-modal regime. In practice, fused multi-modal tomography is performed in the regime with equal or more HAADF projections than chemical (i.e., top-left triangle). Multimodal tomography also performs well when the chemical projections exceed the number of HAADF projections, however, this is not practical since HAADF signals can be acquired simultaneously with EDX and EELS. Similar trends are observed in a second large-scale simulation performed on a synthetic composite structure composed of transition metal CoO nanoparticles embedded in a NiO support (Supplementary Fig. 10).

Fused multi-modal electron tomography can measure 3D stoichiometry without knowledge of inelastic cross-sections. The ratio of chemical concentrations for each voxel quantifies local stoichiometry. For the simulated CoO-CuO nanocubes, values agree with the ground truth (Supplementary Fig. 11)–concentrations of Cu, Co, and O are centered at the expected value of 0.50. Here, stoichiometric precision of multi-modal tomography ( $\sigma$ =0.04) is four times better than traditional chemical tomography ( $\sigma$ =0.17). For experimental Fe<sub>3</sub>O<sub>4</sub> nanoparticles (Supplementary Fig. 12), multi-modal tomography produces an average Fe concentration of 0.46 (0.43 expected) with a standard deviation of 0.15. Note, determining stoichiometry using traditional

chemical tomography also requires accurate calculation of the inelastic cross-sections for each experiment<sup>48</sup>.

#### Discussion

Although this work presents significant advantages for fused multimodal electron tomography, the technique requires supervision. Convergence should be verified and reasonable weights for each term in the cost function (Eq. (1)) should be assigned (Supplementary Fig. 13). Error when implementing routine spectroscopic preprocessing (e.g., incorrect background subtraction of EELS spectra49 or failure to decouple overlapping characteristic X-ray peaks) will cause inaccurate stoichiometric quantification. These errors will amplify when applied to multi-modal data fusion. Although lighter elements have smaller elastic cross-sections, they tend to have larger inelastic cross-sections which benefits chemical tomography. For example, the K-shell cross-section (chemical spectroscopic signal) of Carbon (Z = 6) is over 20-fold larger than Germanium (Z = 32)<sup>50,51</sup>. EELS is advantageous for discerning lighter elements whereas overlapping peaks may occur in EDX. In general, electron tomography is favorable for measuring volumes in the range of (10 nm)<sup>3</sup> to (1000 nm)<sup>3</sup> at resolutions around 3 to 30 Å<sup>4</sup>. Thick specimens with dimensions that far exceed the mean free path of the electron can produce inversion contrast that will cause electron tomography to fail<sup>52</sup>-also causing failure for multi-modal electron tomography (Supplementary Fig. 14). Electron tomography performs best for thicknesses less than three times the incident electron's mean free path (e.g., < 550 nm for Silicon at 300 keV)<sup>53</sup>. In all electron tomography experiments, beam convergence angles should be chosen to match the desired resolution and depth of focus<sup>41</sup>. As shown for 2D fused multi-modal electron microscopy27, fused multi-modal tomography works best when elements have discernible contributions to the HAADF contrast and all chemical elements have been imaged. Multi-modal tomography leverages CS (e.g., TV min.) which assumes incoherence (i.e., a high level of dissimilarity) between the sensing and sparsifying transform<sup>54-56</sup>—although this assumption typically holds as demonstrated for the datasets presented herein.

In summary, we present fused multi-modal electron tomography that enables chemically-sensitive 3D reconstruction of matter with nanometer resolution at high SNR. We demonstrate that researchers do not need to choose between measuring 3D structure without chemical detail or characterizing chemistry along a single viewing direction. By linking signals from elastic (HAADF) and inelastic (EDX/EELS) scattering processes, the traditional dose limits of chemical tomography are substantially surpassed. In some cases, a one-hundred-fold reduction in dose is estimated. When compared to 2D chemical imaging<sup>27</sup>, we show the benefits of data fusion in 3D are much greater. To demonstrate, the complete volumetric density of each chemistry was mapped in several systems including Au-Fe<sub>3</sub>O<sub>4</sub>, Co<sub>3</sub>O<sub>4</sub>-Mn<sub>3</sub>O<sub>4</sub>, ZnS-Cu<sub>0.64</sub>S<sub>0.36</sub>, and Cu-SiC nanomaterials. In both artificial and experimental datasets, fused multi-modal electron tomography shows significant benefits in the accuracy of 3D chemical imaging. This approach enables chemical tomography of a wide range of previously inaccessible materials with moderate radiation sensitivity. At chemical resolutions of 1nm, fused multi-modal electron tomography will facilitate understanding of geometrically complex materials-from 3D semiconductor gate stacks57, clean energy materials58,59, or photoluminescence quantum dot nanoparticles<sup>60</sup>.

Here, fused multi-modal tomography used commonly available STEM detectors (HAADF, EDX, and EELS), however, future approaches can further integrate other modalities—such as 4D-STEM from pixelarray detectors<sup>61</sup>, annular bright field<sup>62</sup>, ptychography<sup>63</sup>, low-loss EELS<sup>64</sup>. Furthermore, the tremendous potential of multi-modal data fusion as a paradigm readily enhances deep learning to capitalize on the unique advantages from both domains<sup>65</sup>.

#### Methods

#### Specimen synthesis and preparation

Au-Fe<sub>3</sub>O<sub>4</sub> superlattice nanoparticles. Syntheses of 3.9 nm Au NPs<sup>66</sup> and 10.2 nm Fe<sub>3</sub>O<sub>4</sub> NPs<sup>67</sup> were carried out under nitrogen atmosphere using standard Schlenk line techniques according to literature methods. Polystyrene-based ligands were attached to the NP surface through a ligand exchange process as reported before<sup>68</sup>. Thiolterminated PS (PS-SH) was used as the polymeric ligand for Au NPs and was synthesized using Radical Addition Fragmentation Transfer polymerization and end-functionalized by aminolysis. Amineterminated polystyrene was used as the polymeric ligand for Fe<sub>3</sub>O<sub>4</sub> NPs and was synthesized using atom transfer radical polymerization and end-group modification<sup>69</sup>. Binary superlattice of Au and Fe<sub>3</sub>O<sub>4</sub> NPs was prepared by nanoparticle co-crystallization at water-air interface. A toluene solution containing two types of NPs with concentration ratio of 2:1 was drop-cast onto the water surface in a Teflon well and slowly dried overnight. The binary nanoparticle film was transferred onto a 200-mesh carbon TEM substrate and further dried in vacuum oven for 6 h to remove residual solvent.

 $Co_3O_4$  nanocubes. A mixture of 0.37 g of cobalt(II) perchlorate (Aldrich) and 2.7 g of oleylamine (Acros) in 15 mL of 1-octanol (Aldrich) was heated to 120 °C under air and aged for 2 h. During the heating, 0.7 mL of distilled water was added before the temperature reaches 120 °C. After the reaction, an excess amount of acetone and ethanol was added and  $Co_3O_4$  nanocubes were retrieved by centrifugation.

**Core-shell Co<sub>3</sub>O<sub>4</sub>-Mn<sub>3</sub>O<sub>4</sub> nanoparticles.** An organic/aqueous suspension was prepared by adding 0.080 g of Co<sub>3</sub>O<sub>4</sub> nanocubes into a mixture of oleylamine (5 mmol), oleic acid (0.5 mmol), formic acid (3.15 mmol, Aldrich), and 15 mL of xylenes (Aldrich). The as-prepared suspension was heated to 40 °C under air and aged for 3 h with magnetic stirring. And then, 0.7 mL of 0.7 M aqueous solution of manganese (II) chloride tetrahydrate was rapidly injected into the suspension at 90 °C and aged for 1.5 h under air. After the reaction, the nanocrystals were washed with hexane/ethanol and retrieved by centrifugation. The final product was prepared with three iterations of this process.

ZnS-Cu<sub>0.64</sub>S<sub>0.36</sub> nanocrystals. Synthesis of the ZnS-Cu<sub>0.64</sub>S<sub>0.36</sub> Heterostructured NPs was performed as described by literature using typical air and water free synthetic techniques<sup>29</sup>. Cu<sub>1.81</sub>S (roxbyite) nanocrystals are synthesized by first dissolving CuCl<sub>2</sub>·2H<sub>2</sub>O in oleylamine (OLAM) at 200 °C after thoroughly degassing the solution at high temperature. Tert-butyl-disulfide is then injected at 180 °C and the reaction continues at this temperature for 40 min. After cooling to room temperature, the NPs are washed with hexanes and acetone then dried in a vacuum desiccator. The roxbyite NPs are then injected into a concentrated Zn ion solution heated at 50 °C for 10 min to facilitate the reaction. Briefly, ZnCl<sup>2</sup> and OLAM are degassed at high temperature and then heated at 180 °C for 30 min to make a concentrated solution of Zn<sup>2+</sup> for cation exchange. After cooling the Zn<sup>2+</sup> solution to 100 °C, an aliquot of the solution is mixed with toluene and the temperature is adjusted to 50 °C. The synthesized roxbyite NPs are dissolved in trioctyl phosphine and then injected into the Zn<sup>2+</sup> solution in and allowed to react for 30 min before quenching the reaction with cold acetone.

**Cu-SiC catalyst**. The Cu/SiC catalyst was prepared on a commercial SiC support purchased from a commercial vendor with a Brunauer-Emmett-Teller surface area of 30 m<sup>2</sup>/g and pore volume of 0.4 cm<sup>3</sup>/g. following previously described methods<sup>70</sup>. The catalyst was prepared by incipient wetness impregnation using a 2M Cu(NO<sub>3</sub>)<sub>2</sub>·3H<sub>2</sub>O aqueous solution followed by drying (120 °C for 2 h) and calcination in air (350 °C for 2 h) at a heating rate of 2 deg/min.

**Acrylic C-TiO<sub>2</sub> nanoparticles.** The C-TiO<sub>2</sub> sample was prepared by blending commercial TiO<sub>2</sub> particles (purchased from Chemours) with an emulsion polymer latex. Before conducting the chemical imaging at room temperature, the blend was pre-treated under the electron beam in a Thermo Fisher T12 TEM at -80 °C to promote cross-linking in the latex and preserve its morphology above the glass transition temperature.

#### **Electron tomography acquisition**

Simultaneously acquired HAADF and EELS tilts series for the Au-Fe<sub>3</sub>O<sub>4</sub> specimen were collected on a Talos F200X G2 (Thermo Fisher) operated at 200 keV with a probe current of 115 pA, probe semi-angle of roughly 10.5 mrad and inner collection semi-angle of 50 mrad. The HAADF projections were collected from  $-60^{\circ}$  to  $+60^{\circ}$  with a 3° angular increment using a Model 2021 Fischione Analytical Tomography Holder. At each tilt angle, a STEM image with a 24 µs dwell time at each pixel of a lateral dimension 6.4 Å. Simultaneously acquired HAADF and EELS spectrums were acquired at acquired with a 15° angular increment with a dwell time of 3 ms receiving a total electron dose of  $4.9 \times 10^5 \text{ e/Å}^2$  ( $1.72 \times 10^4 \text{ e/Å}^2$ ,  $4.73 \times 10^5 \text{ e/Å}^2$  for the HAADF and EELS modality, respectively). Refer to Supplementary Figs. 15, 16 to view the raw tilt series.

Simultaneously acquired HAADF and EELS tilt series for the Co<sub>3</sub>O<sub>4</sub>-Mn<sub>3</sub>O<sub>4</sub> specimen were collected on a double aberrationcorrected modified FEI Titan 80-300 microscope (the TEAM I instrument at the National Center for Electron Microscopy within Lawrence Berkeley National Laboratory) operated at 300 keV with a probe current of 115 pA and semi-angle of roughly 10 mrad. This microscope is equipped with a Gatan K3 detector and Continuum spectrometer. The HAADF projections were recorded from -60° to +60° with a 3° angular increment using a Hummingbird Scientific eucentric Tomography Holder. At each tilt angle, a STEM image with a 24 µs dwell time at each pixel of a lateral dimension of 7.79 Å. Simultaneously acquired HAADF and EELS spectrums were acquired at acquired with a 15 ° angular increment with a dwell time of 0.677 ms receiving a total electron dose of  $8.37 \times 10^4 \text{ e/Å}^2$  (1.16  $\times 10^4 \text{ e/Å}^2$ , 7.21  $\times 10^4 \text{ e/Å}^2$  for the HAADF and EELS modality, respectively). Refer to Supplementary Figs. 17, 19 to view the raw tilt series.

Simultaneously acquired HAADF and EDX tilt series for the Cu-SiC specimen were collected on a Talos F200X G2 (Thermo Fisher) operated at 200 keV with a probe current of 250 pA, probe semi-angle of roughly 10.5 mrad and collection angle of 44–200 mrad. The HAADF projections were collected from –75 to +70 with a 3° angular increment. At each tilt angle, a STEM image with a 20 µs at each pixel of the lateral dimension of 1.4679 nm. Simultaneously acquired HAADF and EDX spectrums were acquired at acquired with a 15° angular increment with a dwell time of 20 µs dwell time for 25 frames receiving a total electron dose of  $4.33 \times 10^4$  e/Å<sup>2</sup> (7.1 × 10<sup>3</sup> e/Å<sup>2</sup>,  $3.62 \times 10^4$  e/Å<sup>2</sup> for the HAADF and EELS modality, respectively). The initial chemical distributions were generated from EDX maps using commercial Velox software that produced initial net count estimates (however atomic percent estimates are also suitable).

#### Multi-modal tilt series alignment

The EELS signals were obtained by integration over the core loss edges, all of which were done after background subtraction. The background EELS spectra were modeled using a linear combination of power laws implemented using the open-source Cornell Spectrum Imager software<sup>9</sup>.

Before tilt series alignment, the spectrum images have been driftcorrected after acquisition assuming a time-dependent linear drift model, as illustrated in Supplementary Fig. 20. The survey image, which is taken with an identical dwell time as the HAADF tilts, is taken as a reference. Iterative image registration between the chemical and HAADF signals seek an optimal translation and affine transformation. For the alignment of the tilt series, a coarse alignment is performed with either the center of mass (CoM) or cross-correlation method<sup>71</sup>. CoM works best when the total projected volume is fixed across specimen tilts (i.e., the object is isolated)<sup>72</sup>. In cases where either of these requirements are not met (e.g., fields of view where multiple particles are visible as demonstrated with the Au-Fe<sub>3</sub>O<sub>4</sub> nanoparticles), cross-correlation should be considered. Fine alignment is performed with custom written projection matching method<sup>73</sup> on the HAADF modality. The measured translation shifts are subsequently applied to the corresponding tilts where simultaneously acquired chemical maps were acquired.

#### Fused multi-modal tomography recovery

Here, fused multi-modal electron microscopy is framed as an inverse problem expressed in the following form  $\hat{x} = \arg \min_{x \ge 0} \lambda_1 \Psi_1(x) + \lambda_2 \Psi_2(x) + \lambda_3 TV(x)$  where  $\hat{x}$  is the final reconstruction, and the three terms are described in the main manuscript (Eq. (1)). When implementing an algorithm to solve this problem, we concatenate the multi-element spectral variable (x) as 2D matrices:  $\boldsymbol{x} \in \mathbb{R}^{n_y \cdot n_y \cdot n_i \times n_x}$  where  $n_i$  denotes the total number of reconstructed elements and  $n_{x_i}$ ,  $n_y$  represent number of pixels in the x and y direction and  $x_i$ ,  $b_i$  are the reconstructions and chemical maps for element *i*  $(\mathbf{x}_i \in \mathbb{R}^{n_y \cdot n_y \times n_x} \text{ and } \mathbf{b}_i \in \mathbb{R}^{n_y \cdot N_{\text{chem}}^{\text{proj}} \times n_x})$ . Here the axis of rotation is along the x-direction  $(n_x)$ .

The optimization problem is solved by a combination of gradient descent with TV regularization. We minimize this cost function by iteratively descending along the negative gradient directions for the first two terms and subsequently evaluate the isotropic TV proximal operator to denoise the chemical volumes<sup>74</sup>. The gradients of the first two terms are:

$$\nabla_{\boldsymbol{x}} \Psi_1(\boldsymbol{x}) = -\gamma \operatorname{diag} \left( \boldsymbol{x}^{\gamma-1} \right) \boldsymbol{\Sigma}^T \boldsymbol{A}_h^T \left( \boldsymbol{A}_h \left( \boldsymbol{\Sigma} \boldsymbol{x}^{\gamma} \right)^T - \boldsymbol{b}_H \right)$$
(2)

$$\nabla_{\boldsymbol{x}_i} \Psi_2(\boldsymbol{x}_i) = \boldsymbol{A}_c^T \left( (\boldsymbol{A}_c \boldsymbol{x}_i - \boldsymbol{b}_i) \oslash (\boldsymbol{A}_c \boldsymbol{x}_i + \boldsymbol{\varepsilon}) \right), \tag{3}$$

where  $\oslash$  denotes point-wise division,  $\boldsymbol{b}_{H} \in \mathbb{R}^{n_{y}N_{\text{HADF}}^{\text{proj}} \times n_{x}}$  are the HAADF measurements,  $\boldsymbol{A}_{h} \in \mathbb{R}^{n_{y}\cdot N_{\text{HADF}}^{\text{proj}} \times n_{y}\cdot n_{y}}$  and  $\boldsymbol{A}_{c} \in \mathbb{R}^{n_{y}\cdot N_{\text{chem}}^{\text{proj}} \times n_{y}\cdot n_{y}}$  are forward projection matrices operating on the chemical and HAADF modalities. Here, the first term in the cost function, relating the elastic and inelastic modalities, has been equivalently re-written as  $\Psi_{1} = \frac{1}{2} \parallel \boldsymbol{A}_{h}(\boldsymbol{\Sigma}\boldsymbol{x}^{y}) - \boldsymbol{b}_{H} \parallel_{2}^{2}$ , where  $\boldsymbol{\Sigma} \in \mathbb{R}^{n_{y}\cdot n_{y}\cdot n_{y}\cdot n_{i}}$  and  $\boldsymbol{\Sigma}\boldsymbol{x}$  expresses the summation of all chemistries as matrix-vector multiplication. Evaluating the TV proximal operator is in itself another iterative algorithm. In addition, we impose a non-negativity constraint since negative concentrations are unrealistic. We initialize the first iterate with reconstructions composed purely of the raw measured data ( $\boldsymbol{x}_{i}^{0} = \arg\min \Psi_{2}$ ). This is an ideal starting point as it is a local minimizer of  $\Psi_{2}$ .

Appropriate step sizes for convergence of Eq. (1) can be determined estimating the Lipschitz constant of the measurement matrix using the Power method<sup>75</sup>. Convergence can be confirmed by assessing each term in the cost function as the reconstruction proceeds (Supplementary Fig. 13). Sub-optimal parameters often result in slower convergence. Smooth and asymptotic decay of all three terms in Eq. (1) is an indicator of reliable reconstruction. We find that the optimal weights ( $\lambda$ ) in Eq. (1) do not change significantly between datasets and even sub-optimal terms outperform traditional tomography methods. However careful selection can also be achieved by selecting values within the inflection point of the Pareto front in Eq. (1) (see Supplementary Fig. 13)<sup>76</sup>. The final 3D HAADF and multi-modal chemical volumes were rendered using the Tomviz platform (tomviz.org<sup>77</sup>).

# Multi-modal simulations and Bayesian hyperparameter optimization

To demonstrate the functionality of our fused multi-modal electron tomography algorithm, we created a multi-channel phantom specimen inspired from an experimental system. The phantom consists of four channels, which we attribute to the crystal stoichiometry of CuO, CoO, and Au (Fig. 4c) with a volume size of 256<sup>3</sup>. The HAADF intensity is proportional to  $\sum_{e}(Z_i x_i)^{v}$  where  $x_i$  reflects the element's stoichiometry. To produce chemical maps with realistic noise characteristics, we set the background (vacuum) to roughly 15% of the max intensity and subsequently applied Poisson noise to meet the desired SNR. For a

Poisson-limited signal, each synthetic image has an SNR of  $\frac{\mu_s + \mu_s^2}{\sigma_z^2}$  where

 $\mu_s$  is the mean signal and  $\sigma_N^2$  is the variance of noise<sup>41</sup> In the case of Fig. 4, the SNR of the Co, Cu, O, Au, and HAADF modalities were 1.92, 2.89, 2.69, 1.96, 2208.67, respectively. Prior to measuring the NRMSE of the reconstructed volumes, the chemical distributions were normalized with zero mean and unit standard deviation. The NRMSE expresses a normalized measure of agreement between the recon-

structed (**x**) and ground truth (**y**) :  $\sqrt{\frac{\sum_{ij,k} (\mathbf{y}_{ij,k} - \mathbf{x}_{ij,k})^2}{\sum_{ij,k} (\mathbf{y}_{ij,k})^2}}$ . While the HAADF

SNR may be high, we found the NRMSE reliably converges when above 50 (Supplementary Fig. 21).

Determining optimal regularization parameters for the phase diagram (Fig. 4a) is computationally expensive to explore due to its variability across sampling conditions. While grid search could find the best parameters by exhaustively exploring all possible candidate values, the computation time would be expensive as each map would take ~125 days to complete on a single GPU.

We efficiently explored the parameter space with Bayesian optimization (BO)–a machine learning framework known for optimizing expensive unknown objective functions with minimal evaluations<sup>78,79</sup>. It works by building a probabilistic model of the objective function with Gaussian processes (GP) regression. GP not only estimates our function of interest but also provides the uncertainty measurements to guide future predictions. BO takes into account past evaluations when determining future hyperparameter selections via an acquisition function<sup>80</sup>. For our simulations, we carried out BO with GP in Python with the Scikit Optimize library (scikit-optimize.github.io/stable) with the Matern kernel and GP Hedge acquisition strategy<sup>81</sup>. By exploiting BO with GP, we are able to provide an atlas of balanced hyperparameters for Eq. (1) with the CoNiO and CoCuO synthetic datasets (Supplementary Figs. 22, 23). The estimated parameter landscape is smooth and continuous with a clear global optimum.

Asynchronous parallel BO on supercomputing resources allowed us to efficiently run several reconstructions simultaneously on a single node. This form of parallel computing resulted in several factors of computational speed up as multiple GPUs received unique experimental parameters (e.g., SNR or sampling) to reconstruct concurrently amongst each other. Specifically, the computation time to generate an NRMSE map was reduced by 99.8%–taking less than a day to complete (18 h). In total, 3452 GPU hours were used to complete these simulations–1078 h on Summit - OLCF and 1078 h on ThetaGPU–ALCF for the phase diagrams (Fig. 4 and Supplementary Fig. 10). An additional 1296 GPU hours on Summit were used to produce the SNR plots (Supplementary Fig. 21).

#### Data availability

The raw and aligned Au-Fe<sub>3</sub>O<sub>4</sub>,  $Co_3O_4$ -Mn<sub>3</sub>O<sub>4</sub>, and Cu-SiC tilt series with reconstructed 3D chemistries are available in a Zenodo repository<sup>82</sup>.

#### **Code availability**

All of the multi-modal electron tomography reconstruction and iterative alignment codes are available at github.com/jtschwar/tomo\_TV and github.com/jtschwar/projection\_refinement<sup>83</sup>. A sample jupyter notebook outlining the fused multi-modal reconstruction on the Cu-SiC and Au-Fe<sub>3</sub>O<sub>4</sub> material systems will be available in the tomo TV repository.

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#### Acknowledgements

R.H. and J.S. acknowledge support from The Dow Chemical Company. R.H. acknowledges support from the U.S. Department of Energy, Basic Energy Sciences, under award DE-SC0024147. This work made use of the Michigan Center for Materials Characterization (MC2) and Molecular Foundry, Lawrence Berkeley National Laboratory. The Molecular Foundry was supported by the Office of Science, Office of Basic Energy Sciences, of the U.S. Department of Energy under Contract No. DE-AC02-05CH11231. M.C. and M.G.C. were funded by the US Department of Energy in the program "4D Camera Distillery: From Massive Electron Microscopy Scattering Data to Useful Information with AI/ML. The authors thank Tao Ma, Bobby Kerns, and Chengyu Song for their assistance at MC2 and NCEM. Y.W.Q. and T.X. acknowledge supports from the U.S. Department of Energy, Office of Science, Office of Basic Energy Sciences, Materials Sciences and Engineering Division, under contract DE-AC02-05CH11231 (Organic–Inorganic Nanocomposites KC3104). R.D.R. acknowledges support from NSF under grant DMR-1809429. This research used the Oak Ridge Leadership Computing Facility at the Oak Ridge National Laboratory and Argonne Leadership Computing Facility at Argonne National Laboratory, which is supported by the Office of Science of the U.S. Department of Energy under Contract No. DE-AC05-00OR22725 and DE-AC02-06CH11357.

## **Author contributions**

J.S., Y.J. and R.H. conceived the idea. J.S., J.P. and R.H. implemented the multi-modal reconstruction algorithms. J.F. and Z.W.D. validated algorithms and provided mathematical rigor. J.S., J.M., H.Z. and R.H. conducted the chemical tomography simulations. J.S. conducted the EELS and EDX tomography experiments designed by J.S., R.H., M.S. and P.E. Y.Q. and T.X. synthesized the Au-Fe<sub>3</sub>O<sub>4</sub> nanoparticles and provided data interpretation. M.G.C. and M.S. synthesized the  $Co_3O_4$ -Mn<sub>3</sub>O<sub>4</sub> coreshell nanoparticles and provided data interpretation. J.R. and R.R. synthesized the ZnS-CuS nanoparticles and provided data interpretation. S.R. and A.K. provided the Cu-SiC nanoparticles and data interpretation. J.S. and R.H. wrote the manuscript. All authors reviewed and edited the manuscript.

## **Competing interests**

The authors declare no competing interests.

## **Additional information**

**Supplementary information** The online version contains supplementary material available at https://doi.org/10.1038/s41467-024-47558-0.

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**Peer review information** *Nature Communications* thanks the anonymous reviewer(s) for their contribution to the peer review of this work.

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# **Supplementary Information for**

# Imaging 3D Chemistry at 1 nm Resolution with Fused Multi-Modal Electron Tomography

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## 1 Multi-Modal Reconstruction of Au-Fe<sub>3</sub>O<sub>4</sub> Nanoparticles Inside a Carbon Support



Fig.1 | The complete reconstruction of the Au-Fe $_3O_4$  nanoparticle superlattice inside the carbon matrix (highlighted in purple). Scale cube, 10 nm<sup>3</sup>.

#### 2 Consequence of Reconstructing CuO/CoO Simulation with Individual Components in Cost Function



**Fig.2** | **a** Synthetic CuO/CoO nanocubes ground truth EELS maps and simultaneous HAADF reconstruction. **b** Slice of the reconstructed nanocubes dataset with only  $\Psi_2$  (data consistency),  $\lambda_1, \lambda_3 = 0$ . The recovered EELS maps and HAADF reconstruction are noisy. **c** Slice of the reconstructed multi-modal nanocubes dataset with  $\Psi_1$  and  $\Psi_2$  (model + data-consistency),  $\lambda_3 = 0$ . The EELS maps are still noisy but show improvement with the extra  $\Psi_1$  term. **d** Slice of the reconstructed multi-modal nanocubes dataset with  $\Psi_2 + TV$ ,  $\lambda_1 = 0$ . The expression  $\Psi_2 + TV$  is equivalent to a denoising problem; thus the resulting reconstructions produce common staircase artifacts associated with TV [1]. **e** Slice of the reconstructed nanocubes dataset with fused multi-modal electron tomography. Scale bar, 75 nm.

#### 3 Consequence of Reconstructing CoNiO Simulation with Individual Components in Cost Function



**Fig.3** | **a** Synthetic CoNiO nanotube ground truth EELS maps and simultaneous HAADF reconstruction. **b** Slice of the reconstructed nanotube dataset with only  $\Psi_2$  (data consistency),  $\lambda_1, \lambda_3 = 0$ . The recovered EELS maps and HAADF reconstruction are noisy. **c** Slice of the reconstructed multi-modal nanotube dataset with  $\Psi_1$  and  $\Psi_2$  (model + data-consistency),  $\lambda_3 = 0$ . The EELS maps are still noisy but show improvement with the extra  $\Psi_1$  term. **d** Slice of the reconstructed multi-modal nanotube dataset with  $\Psi_2$  + TV,  $\lambda_1 = 0$ . The expression  $\Psi_2$  + TV is equivalent to a denoising problem; thus the resulting reconstructions produce common staircase artifacts associated with TV [1]. **e** Slice of the reconstructed nanotube dataset with fused multi-modal electron tomography. Scale bar, 50 nm.

4 Multi-Modal EELS Tomography of ZnS - Cu<sub>0.64</sub>S<sub>0.36</sub> Heterostructured Nanoparticles



Fig.4 | a HAADF tomography on heterostructured nanocrystals with applications in photovoltaic devices and battery electrodes [2]. The copper sulfide properties are sensitive to stoichiometry and crystal structure at the interface. The HAADF reconstruction and 2D slice are shown on the left. b The fused multi-modal reconstruction illustrating CuS or ZnS-rich nanoparticles and oxidized shells. c 2D slices of the chemical reconstructions with the noisy traditional reconstructions highlighted on the left of each image. Scale bar, 50 nm. d The HAADF and fused multi-modal chemical tomogram for a smaller field of view. e Representative EELS spectra for the S, C, Cu, and Zn core loss edges.

The simultaneously acquired HAADF and EELS tilt series for the specimen were collected on a Talos F200X G2 (Thermo Fisher) operated at 200 keV with a probe semi-angle of roughly 10.5 mrad and inner collection semi-angle of 50 mrad. For the larger cluster (**a**-**c**), the HAADF projections were collected from -74° to +70° with a 3° angular increment using a Model 2021 Fischione Analytical Tomography Holder. At each tilt angle, a STEM image with a  $\mu$ s dwell time at each pixel of a lateral dimension of 1.26 nm. Simultaneously acquired HAADF and EELS spectrums were acquired at acquired with a 15° angular increment with a dwell time of 2 ms receiving a total electron dose of  $8.25 \times 10^4 \ e/Å^2$  ( $3.87 \times 10^3$ ,  $7.86 \times 10^4$  for the HAADF and EELS modality, respectively).

For the smaller cluster (d), the HAADF projections were collected from  $-74^{\circ}$  to  $+67^{\circ}$  with a 3° angular increment using a Model 2021 Fischione Analytical Tomography Holder. At each tilt angle, a STEM image with a 30  $\mu$ s dwell time at each pixel of a lateral dimension of 1.12 nm. Simultaneously acquired HAADF and EELS spectrums were acquired at acquired with a 15° angular increment with a dwell time of 3.5 ms receiving a total electron dose of  $2.07 \times 10^5 \ e/Å^2$  ( $6.87 \times 10^3$ ,  $2.01 \times 10^5$  for the HAADF and EELS modality, respectively).

## 5 Multi-Modal EDX Tomography of Cu-SiC Nanoparticles



Fig.5 | a Cu-supported silicon carbon (SiC) catalysts designed for the production of fuels and chemicals from biomass. These silica supported catalysts efficiently convert ethanol into acetaldehyde because of their high selectivity and stability [3]. The HAADF reconstruction with a few tilt micrographs is shown on right. b The fused multi-modal reconstruction highlighting Cu nanoparticles embedded inside the SiC support and raw EDX maps are shown on the right. Scale bar, 50 nm. c The PSF of an individual 3 nm nanoparticle inside the SiC. A few 2D slices of the reconstruction are shown on the right where we see the structure is sharp along the orthogonal axis perpendicular to the missing wedge and approximately a 20% reduction in resolution along the missing wedge direction. Scale bar, 2 nm. d EDX spectra for a single tilt.

## 6 Historical Demonstrations of Multi-Element Chemical Tomography



Fig.6 The reported dose and Nyquist limited resolutions for the fused multi-modal (MM) reconstructions reported in this manuscript are compared to previous multielement chemical tomography (CT) experiments [4–10]. Note, the actual achieved 3D resolution of previously reported chemical tomography may be lower than the Nyquist resolution.

#### 7 Resolution and Electron Dose for Multi-Modal Electron Tomography



Fig.7 | a Resolution and dose relationship for electron tomography approximates the best achievable resolution for each material [11–15] – assuming an image contrast of 80%[16]. Fused multi-modal electron tomography results in a much higher dose-efficiency which enables higher resolution at any dose limit. b Multi-modal (green) and conventional chemical tomography (blue). This relationship between dose and resolution assumes sufficient tomographic sampling is achieved (i.e. many projections)— in practice the actual resolution will be much lower. Dose limited resolution assumes the material is adequately sampled (i.e. Crowther and Nyquist relations)

## 8 Measuring the Resolution of Au-Fe<sub>3</sub>O<sub>4</sub> Nanoparticles in the HAADF



**Fig.8** | **a** Fused EELS tomograms of Au-Fe<sub>3</sub>O<sub>4</sub> nanoparticles. Power spectral density of the HAADF reconstruction along the principal axial directions shown on the right. Scale cube, 2 nm<sup>3</sup>. Scale bar, 0.5 nm<sup>-1</sup>. **b** Power spectral density profiles for YZ and XY planes. **c** Line scan profiles of a 2.5 nm Au nanoparticle give a resolution of 1.00, 1.03, and 1.01 nm along the x, y, and z directions.

#### 9 Sampling Requirements of Multi-Modal Tomography on CoO/CuO Nanocubes



Fig.9 | a An NRMSE map representing reconstruction error as a function of the number of HAADF and chemical tilts. The bottom two rows provide results from non-multimodal algorithms where the HAADF is not included in the reconstruction process. The last row uses the Poisson Maximum Likelihood term to reconstruct the individual chemical distributions. The second to last row adapts the compressed sensing framework into the reconstruction process by including total variation minimization – thus providing a 2-3 fold reduction in average error. **b** Visualization of four points in the phase diagram corresponding to conventional chemical tomography, regularized tomography (compressed sensing), and low or high-dose fused multi-modal electron tomography. The 3D models were then rendered and colored in Tomviz [17]. Scale bar, 75 nm.

#### 10 Estimating the Sampling Requirements for a CoNiO Composite



**Fig.10** | **a** A normalized root-mean-square-error (NRMSE) heatmap of a fused multi-modal CoNiO nanotube reconstructions as a function of the number of HAADF and chemical tomographic projections. Brighter pixels denote higher levels of error in the reconstruction when compared to the ground truth. The SNR for the Co, Ni, O, and HAADF modalities were 2.66, 6.46, 3.17, and 2156.16 respectively. **b** Visualization of four points in the phase diagram corresponding to conventional chemical tomography, regularized tomography (compressed sensing), and low or high-dose fused multi-modal electron tomography.**c** The synthetic CoNiO nanotube ground truth 3D models generating synthetic chemical and ADF projections. The 3D models were then rendered and colored in Tomviz [17]. Scale bar, 50 nm.



**Fig.11** | Histograms of the chemical concentrations for each voxel in the traditional and fused multi-modal tomography reconstructions are shown for the simulated CuO-CoO nanocube system. The mean values of each chemistry are within  $\pm 0.03$  of the expected value of 0.5. For traditional chemical tomography, the accuracy improves as SNR increases or more projections are collected. Multi-modal tomography maintains low error, especially for experimentally realistic conditions (e.g. 14 chemical tilts and SNR < 10).

# Traditional Chemical Tomography

Multi-Modal Chemical Tomography



Fig.12 | 3D chemical reconstructions for each element are shown with their corresponding voxel intensity histograms. The mean values and standard deviations are  $46.4\pm15.1\%$ ,  $54.6\pm15.3\%$ ,  $100\pm0\%$  for Fe, O, and Au, respectively. The expected stochiometry of this system is 42.9%, 57.1%, 100%.

## 13 Assessing Convergence and Selecting Hyperparameters with Paraeto Front Curves



**Fig.13** | a Pareto fronts illustrates the relationship between reconstruction quality and regularization parameters for multi-modal electron tomography. Depicted are the tradeoffs from three reconstruction evaluation metrics: the multi-modal, self-consistency and average NRMSE across all elements. We see the highest quality reconstruction (lowest NRMSE) occurs around the inflection point of the pareto front. **b** The three individual components in the cost function plotted throughout the multi-modal electron tomography reconstruction process illustrates smooth asymptotic convergence. Convergence should be confirmed for accurate reconstruction.

#### 14 Consequence of Reconstructing Thick 280 nm Nanoparticles with EELS Tomography



Fig.14 | a The HAADF reconstruction with a few HAADF projections from the tilt series shown on right. These projections were square rooted to visually see the TiO and C nanoparticles. b The fused multi-modal reconstruction illustrating TiO nanoparticle decorated by C support with raw EELS maps shown on the right. Scale bar, 100 nm.

The simultaneously acquired HAADF and EELS tilt series for the specimen were collected on a Talos F200X G2 (Thermo Fisher) operated at 200 keV with a probe semi-angle of roughly 10.5 mrad and inner collection semi-angle of 50 mrad. The HAADF projections were collected from -70° to +70° with a +2° angular increment using a Model 2021 Fischione Analytical Tomography Holder. At each tilt angle, a STEM image with a 32  $\mu$ s dwell time at each pixel of a lateral dimension of 2.015 nm. Simultaneously acquired HAADF and EELS spectrums were acquired at acquired with a 10° angular increment with a dwell time of 2 ms receiving a total electron dose of  $4.96 \times 10^4 \ e/Å^2$  ( $3.48 \times 10^3$ ,  $4.61 \times 10^4$  for the HAADF and EELS modality, respectively)

# 15 Raw HAADF Tilt Series for the Au-Fe $_3$ O $_4$ Nanoparticles

$\theta = -60^{\circ}$	$\Theta = -5\mathcal{P}$	θ = -54°	θ = -51°	$\Theta = -48^{\circ}$	θ = -45
$\theta = -42^{\circ}$	$\theta = -39^{\circ}$	$\theta=-36^{\circ}$	$\theta = -33^{\circ}$	$\theta = -30^\circ$	$\theta = -27^{\circ}$
$\theta = -24^{\circ}$	$\theta = -21^{\circ}$	$\theta=-18^\circ$	$\theta = -15^{\circ}$	$\theta = -12^\circ$	θ = -9°
θ = -6°	$\Theta = -3^{\circ}$	$\theta = 0^{\circ}$	θ = +3	θ = +6	θ = +9*
$\theta = \pm 12^{\circ}$	$\theta=\pm 15^\circ$	$\theta = \pm 18^{\circ}$	$\theta = +21^{\circ}$	$\theta = +24^{\circ}$	$\theta=+2\mathcal{T}$
θ = +30°	θ = +33*	θ = +36°	θ = #39	$\theta=\pm 42^{\rm s}$	θ = +45°
$\theta = +48^{\circ}$	θ = +51°	$\theta=+54^{\rm e}$	$\theta = +57^{\circ}$	θ = +60°	
iĝ ⊨ 0ª Betore Experiment	θ = 0° Afte⊧ Experiment				

**Fig.15** | The 45 projection images with a tilt range from  $-60^{\circ}$  to  $+60^{\circ}$  (shown at the top left of each panel) were measured using ADF-STEM. The total electron dose of the tilt series is  $1.72 \times 10^4$  e/Å<sup>2</sup>. Scale bar shown at the bottom right, 25 nm.

## 16 Raw EELS Tilt Series for the Au-Fe<sub>3</sub>O<sub>4</sub> Nanoparticles



**Fig.16** [The 9 EELS maps with a tilt range from  $-60^{\circ}$  to  $+60^{\circ}$  (shown at the top left of each map) were measured using EELS spectroscopy. The total electron dose of the tilt series is  $4.73 \times 10^5$  e/Å<sup>2</sup>. Scale bar is shown at the bottom right, 50 nm.

17 Raw HAADF Tilt Series for the  $Co_3O_4$  -  $Mn_3O_4$  Core-Shell Nanoparticles



**Fig.17** | The 45 projection images with a tilt range from  $-60^{\circ}$  to  $+60^{\circ}$  (shown at the top left of each panel) were measured using ADF-STEM. The total electron dose of the tilt series is  $1.16 \times 10^4$  e/Å<sup>2</sup>. Scale bar is shown at the bottom right, 50 nm.

## 18 Raw EELS Tilt Series for the $Co_3O_4$ - $Mn_3O_4$ Core-Shell Nanoparticles



**Fig.18** | The 9 EELS maps with a tilt range from  $-60^{\circ}$  to  $+60^{\circ}$  (shown at the top left of each map) were measured using EELS spectroscopy. The total electron dose of the tilt series is  $7.21 \times 10^4$  e/Å<sup>2</sup>. Scale bar is shown at the bottom right, 50 nm.

## 19 Orthogonal Views of Au - $Fe_3O_4$ and $Co_3O_4$ - $Mn_3O_4$ Nanoparticles

Au - Fe<sub>3</sub>O<sub>4</sub> Nanoparticle Superlattice



**Fig.19** [3D renderings of the recovered fused chemistries for the Au-Fe $_3O_4$  nanoparticles (top row) and  $Co_3O_4$  -  $Mn_3O_4$  nanoparticles (bottom row) with the yz and xz plane cut-view images displayed alongside the tomograms. Scale bars, 25 nm and 50 nm respectively.



Fig.20 | Drift correction of the spectrum images: a survey image I(x,y), recorded before the spectrum image acquisition, is used as a reference to calculate the affine transformation on the acquired spectrum map. These parameters are subsequently used to correct for drift in the acquired image and all associated spectroscopic signals.

#### 21 SNR Dependency for Successful Fused Multi-Modal Recovery



**Fig.21** | **a** The initial corrupted chemical distributions for oxygen in the CuO-CoO synthetic dataset with increasing SNR. We see as we exceed the Rose criterion (SNR  $\simeq$  5) internal pores and fine features are visible. **b** A heat map expressing the relationship between average reconstruction error and SNR for either modality (HAADF or Chemistry) when 11 chemical maps ( $\Delta \theta = +1^{\circ}$ ) and 141 HAADF projections ( $\Delta \theta = +1^{\circ}$ ) are available. **c** SNR plot highlighting the average NRMSE as a function of chemical SNR for reconstructions without any regularization or fusion (traditional tomography), without data-fusion (regularized tomography) and within the multi-modal framework. We see substantial improvements in reconstruction quality as we incorporate more terms in the cost function. Most notably, data fusion converges quite rapidly when the SNR is above 6. Visualization of the oxygen elemental reconstruction of the 2D slices from the reconstruction algorithm, without data-fusion with the FISTA reconstruction algorithm and within the multi-modal framework [18, 19]. When the sampling between the chemistry and HAADF is equivalent and chemical SNR is larger, regularized tomography outperforms multi-modal tomography. Visualization of the cost from the reconstruction of the 2D slices from the reconstruction of the 2D slices from the reconstruction of the 2D slices from the reconstruction algorithm and within the multi-modal tomography. Visualization of the cobalt and copper elemental reconstruction of the 2D slices from the reconstruction of the three curves.



#### 22 Hyperparameter Estimation with Bayesian Optimization for the CuO-CoO Nanocubes

**Fig.22** | a Bayesian optimization optimizes the data fusion cost function (shown above) when provided a given number of chemical and HAADF tilts. We see the weights between HAADF and chemical modality ( $\lambda_1$  and  $\lambda_2$  respectively) can vary depending on our certainty in either modality. When the number of tilts is low, the corresponding  $\lambda$  value would decrease and vice-versa. Overall, these maps can guide future scientists to produce multi-modal reconstructions with reasonable hyperparameter selections. **b** 3D visualization of the ground truth Au decorated CuO/CoO nanocubes. Scale bar, 75 nm. **c** Bayesian optimization parameter selection landscape where each black dot represents one of the many attempts to find the minimum NRMSE.

## 23 Hyperparameter Estimation with Bayesian Optimization for CoNiO Composite



**Fig.23** | a Bayesian optimization optimizes the data fusion cost function (shown above) when provided a given number of chemical and HAADF tilts. We see the weights between HAADF and chemical modality ( $\lambda_1$  and  $\lambda_2$  respectively) can vary depending on our certainty in either modality. When the number of tilts is low, the corresponding  $\lambda$  value would decrease and vice-versa. Overall, these maps can guide future scientists to produce multi-modal reconstructions with reasonable hyperparameter selections. **b** 3D visualization of the ground truth CoNiO composite structure. Scale bar, 75 nm. **c** Bayesian optimization parameter selection landscape where each black dot represents one of the many attempts to find the minimum NRMSE.

#### 24 Fusion Multi-Modal Electron Tomography Pseudo Code

Algorithm 1 Fused Multi-Modal 3D Chemical Tomography

1: Objective Function:  $\Psi(\boldsymbol{x}) = \frac{\lambda_1}{2} \|\boldsymbol{A}_h \sum_i (Z_i \boldsymbol{x}_i)^{\gamma} - \boldsymbol{b}_h\|_2^2 + \lambda_2 \sum_i \left( \mathbf{1}^T \boldsymbol{A}_c \boldsymbol{x}_i - \boldsymbol{b}_i^T \log(\boldsymbol{A}_c \boldsymbol{x}_i + \varepsilon) \right) + \lambda_{\text{TV}} \sum_i \|\boldsymbol{x}_i\|_{\text{TV}}$ 2: 3: Input:  $\boldsymbol{b}_h \in \mathbb{R}^{n_y \cdot N_{ ext{HAADF}}^{ ext{proj}} imes n_x}$ ▷ HAADF Tilt Series 4: Input:  $\boldsymbol{b}_c \in \mathbb{R}^{n_i \cdot n_y \cdot N_{ ext{chem}}^{ ext{proj}} imes n_x}$ ▷ Chemical Tilt Series 5: Output:  $x \in \mathbb{R}^{n_i \cdot n_y \cdot n_y imes n_x}$ Reconstructed Chemical Tomogram 6: 7:  $N_{\text{iter}} = 100, \varepsilon = 0.1$ 8:  $\boldsymbol{x}_i^0 = \arg\min_{\boldsymbol{x}} \Psi_2(\boldsymbol{0})$ > Initialize first iterate with reconstruction from raw chemical maps 9: 10: for  $k = 1, N_{iter}$  do ▷ Main Loop for  $s = 1, n_x$  do > Apply the Gradient for Each Slice Along the Axis Parallel to Tilt Axis 11:  $\boldsymbol{x}_{s}^{k} = \boldsymbol{x}_{s}^{k-1} - (\lambda_{1} \nabla_{\boldsymbol{x}} \Psi_{1}(\boldsymbol{x}_{s}^{k-1}) + \lambda_{2} \nabla_{\boldsymbol{x}} \Psi_{2}(\boldsymbol{x}_{s}^{k-1}))$ 12: end for 13: for  $i = 1, n_i$  do 14:  $\boldsymbol{x}_{i}^{k} = \text{tv}_{fgp}_{3D}(\boldsymbol{x}_{i}^{k}, \lambda_{TV})$ Channel-Wise TV Minimization 15: end for  $16^{-1}$ 17: end for 18: return x 19: 20: Comments: 21:  $\nabla_{\boldsymbol{x}} \Psi_1(\boldsymbol{x}) = -\gamma \operatorname{diag}(\boldsymbol{x}^{\gamma-1}) \boldsymbol{\Sigma}^T \boldsymbol{A}_h^T \Big( \boldsymbol{A}_h(\boldsymbol{\Sigma} \boldsymbol{x}^{\gamma}) - \boldsymbol{b}_h \Big)$ 22:  $\boldsymbol{\Sigma} \in \mathbb{R}^{n_y \cdot n_y \times n_y \cdot n_y \cdot n_i}$  expresses the summation operation as matrix vector multiplication. 23:  $\nabla_{\boldsymbol{x}_i} \Psi_2(\boldsymbol{x}_i) = \boldsymbol{A}_c^T \Big( (\boldsymbol{A}_c \boldsymbol{x}_i - \boldsymbol{b}_i) \oslash (\boldsymbol{A}_c \boldsymbol{x}_i + \varepsilon) \Big)$ 24: ⊘ is element-wise division. 25:  $\boldsymbol{x}_i \in \mathbb{R}^{n_y \cdot n_y \times n_x}$  is the reconstruction for element *i*. 26:  $\boldsymbol{b}_i \in \mathbb{R}^{n_y \cdot N_{\text{chem}}^{\text{proj}} \times n_x}$  is the chemical tilt series for element *i*. 27:  $\boldsymbol{x}_s \in \mathbb{R}^{n_y \cdot n_y \cdot n_i}$ .

#### Algorithm 2 3D Gradient Projection (GP) Method (DOI: 10.1109/TIP.2009.2028250)

1: Input:  $b \in \mathbb{R}^{n_x \times n_y} \to 3D$  Volume,  $\lambda \to \text{Regularization Parameter}$ ,  $ng \to \text{Number of Iterations}$ 2: **Output:**  $x^* \rightarrow$  Optimal Solution 3: 4: **function** TV\_GP\_3D( $\mathbf{x}, \lambda, ng$ )  $\boldsymbol{p}_x^0 = \boldsymbol{0} \in \mathbb{R}^{(m-1) imes n imes k}, \quad \boldsymbol{p}_y^0 = \boldsymbol{0} \in \mathbb{R}^{m imes (n-1) imes k}, \quad \boldsymbol{p}_z^0 = \boldsymbol{0} \in \mathbb{R}^{m imes n imes (k-1)}$ 5: for k = 1, ng do ⊳ Main Loop 6:  $(m{p}_x^k,m{p}_y^k,m{p}_z^k) = P_{\mathcal{P}}\Big[(m{p}_x^{k-1},m{p}_y^{k-1},m{p}_z^{k-1}) + rac{1}{26\lambda}\mathcal{L}^Tig(P_C[m{b}-\lambda\mathcal{L}(m{p}_x^{k-1},m{p}_y^{k-1},m{p}_z^{k-1})]ig)\Big]$ 7: end for 8: return  $\boldsymbol{x}^* = P_C[\boldsymbol{b} - \lambda \mathcal{L}(\boldsymbol{p}_x^{ng}, \boldsymbol{p}_u^{ng}, \boldsymbol{p}_z^{k-1}))]$ 9: 10: end function 11: 12: function  $\mathcal{L}(p,q,r)$ ▷ Linear Operation for  $i = 1, ..., n_x$  do 13: for  $j = 1, ..., n_u$  do 14: for  $k = 1, ..., n_k$  do 15: **if** i == 0 or  $i == n_x$  **do**  $p_{i,i,k} = 0$ 16: 17: **if** j == 0 or  $j == n_y$  **do**  $q_{i,j,k} = 0$ **if** k == 0 or  $k == n_k$  **do**  $r_{i,j,k} = 0$ 18:  $\mathcal{L}(\boldsymbol{p}, \boldsymbol{q}, \boldsymbol{r})_{i,j,k} = p_{i,j,k} + q_{i,j,k} + r_{i,j,k} - p_{i-1,j,k} - q_{i,j-1,k} - r_{i,j,k-1}$ 19: end for  $20^{\circ}$ end for 21: 22: end for return  $\mathcal{L}(\boldsymbol{p}, \boldsymbol{q}, \boldsymbol{r})$ 23: 24: end function 25: function  $\mathcal{L}^T(\mathbf{x})$ 26: for  $i = 1, ..., n_x - 1$  do 27: for  $j = 1, ..., n_n - 1$  do 28: for  $k = 1, ..., n_z - 1$  do 29: 30:  $p_{i,j} = x_{i,j} - x_{i+1,j,k}$ 31:  $q_{i,j} = x_{i,j} - x_{i,j+1,k}$ 32:  $r_{i,j} = x_{i,j} - x_{i,j,k+1}$ end for 33: end for 34: 35: end for return  $(\boldsymbol{p}, \boldsymbol{q}, \boldsymbol{r})$ 36: 37: end function 38:

Algorithm 3 Projection and Proximal Functions for the 3D Gradient Projection Method (DOI: 10.1109/TIP.2009.2028250)

1: function  $P_C(x)$ > Orthogonal Projection Operator Onto Convex Set (Non-negativity) return  $\max\{0, \boldsymbol{x}\}$ 2: 3: end function 4: 5: function  $P_{\mathcal{P}}(\boldsymbol{p}, \boldsymbol{q}, \boldsymbol{r})$ ▷ Projection Operator for Isotropic TV Norm for  $i = 1, ..., n_x - 1$  do 6: for  $j = 1, ..., n_y - 1$  do 7:  $\begin{aligned} \mathbf{for} \ k &= 1, \dots, n_z - 1 \ \mathbf{do} \\ denom &= \sqrt{p_{i,j,k}^2 + q_{i,j,k}^2 + r_{i,j,k}^2} \\ p_{i,j,k} &= \frac{p_{i,j,k}}{\{\max\{1,denom\}} \\ q_{i,j,k} &= \frac{q_{i,j,k}}{\max\{1,denom\}} \\ r_{i,j,k} &= \frac{r_{i,j,k}}{\max\{1,denom\}} \end{aligned}$ 8: 9: 10: 11: 12: end for 13: end for 14: end for 15: return (p, q, r)16: 17: end function 18: function  $P_{\mathcal{P}}(\boldsymbol{p}, \boldsymbol{q}, \boldsymbol{r})$ > Projection Operator for Anisotropic TV Norm 19: for  $i = 1, ..., n_x - 1$  do 20: for  $j = 1, ..., n_y - 1$  do 21: for  $k = 1, ..., n_z - 1$  do 22:  $\begin{array}{l} p_{i,j,k} = \frac{p_{i,j,k}}{\max\{1, |p_{i,j,k}|\}} \\ q_{i,j,k} = \frac{q_{i,j,k}}{\max\{1, |q_{i,j,k}|\}} \\ r_{i,j,k} = \frac{r_{i,j,k}}{\max\{1, |r_{i,j,k}|\}} \end{array}$ 23: 24: 25: end for 26: end for 27: end for 28: 29: return  $(\boldsymbol{p}, \boldsymbol{q}, \boldsymbol{r})$ 30: end function

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